Enolexo Aldol Reactions



Direct Catalytic Asymmetric *Enolexo* Aldolizations**

Chandrakala Pidathala, Linh Hoang, Nicola Vignola, and Benjamin List*

The aldol reaction is an exceptionally useful strategic C-C bond-forming reaction for the stereoselective construction of cyclic and acyclic molecules. As a result, several catalytic

[*] Dr. B. List, C. Pidathala, L. Hoang, N. Vignola The Scripps Research Institute Department of Molecular Biology 10550 North Torrey Pines Road, La Jolla, CA 92037 (USA) Fax: (+1) 858-784-7028 E-mail: blist@scripps.edu

[**] Generous support by the NIH (RO1 GM-63914) is most gratefully acknowledged. The cover picture was designed by Mike Pique (Scripps Institute) using a proline image attributed to K. N. Houk and S. Bahmanyar (UCLA). Chicken artwork courtesy of Sheri Hoeger, www.madstencilist.com © 2000.

Zuschriften

asymmetric intermolecular variants, both indirectly, with preformed enolate equivalents, and directly involving unmodified carbonyl compounds have been described. Remarkably, however, there is still only one catalytic asymmetric intramolecular aldol reaction, the proline-catalyzed Hajos–Parrish–Eder–Sauer–Wiechert reaction. While the usefulness of this process has been illustrated in a broad context, only 6-enolendo aldolizations [Eq. (1), n=1] have been described so far. Direct catalytic asymmetric enolexo aldolizations [Eq. (2)] are unknown. Herein we describe the first and highly enantioselective examples of this process.

6-Enolendo aldolizations are very common and favored according to the Baldwin rules.^[4,5] The only catalytic asymmetric variant of this process, the Hajos-Parrish-Eder-Sauer-Wiechert reaction has not been extended to different ring sizes, nor have any proline-catalyzed *enolexo* aldolizations [Eq. (2)] been described.^[6] Although Baldwin-favored in the formation of 3-7 membered rings, *enolexo* aldolizations

are less studied^[7] and direct catalytic asymmetric variants are unknown.^[8]

Recently, we discovered the first proline-catalyzed asymmetric intermolecular aldol reactions[9-11] and proposed a unified one-proline enamine catalysis mechanism of both inter- and intramolecular aldol reactions.[12-14] By inspecting possible transition-state models, we realized that in addition to the established 6-enolendo aldolizations via transition state A, proline should also catalyze corresponding 6-enolexo aldolizations via the chairlike assembly B. Such reactions should provide a highly stereoselective pathway to useful *trans*-1,2-disubstituted cyclohexanes.

Indeed, the reaction of heptanedial^[15] (1a) with a catalytic amount of (S)-proline in dichloromethane gave aldol 2a in high yield and diastereoselectivity, and with excellent enantioselectivity [Eq. (3)].^[16,17]

With this encouragement, we decided to study the scope of this

proline-catalyzed 6-*enolexo* aldolization. Various pentane-1,5-dialdehydes (pimelaldehydes; **1**) were prepared^[18] and then treated with a catalytic amount of (S)- or (R)-proline in dichloromethane [Eq. (4), Table 1]. Both **1b** and **1c** provided

the corresponding cyclic aldols **2b** and **2c** in high yields and excellent diastereo- and enantioselectivities. Surprisingly, we found that a single substituent in the 4-position has an unfavorable effect on the stereoselectivity of the cycloaldo-

Table 1: Proline-catalyzed *enolexo* aldolizations of dicarbonyl compounds. Yields refer to diols obtained after in situ NaBH₄ reduction.

Dicarbonyl	Yield [%]	Products	ee [%]	d.r. [%]
онс сно	95	OHC,,, OH	99	10:1
OHC Me CHO	74	OHC,, Me Me 2b	98	> 20:1
OHC CHO Me Me 1c	75	OHC,, Me Me	97	> 20:1
OHC CHO	76	OHC Me	75,89,95,8	22:5:5:1
OHC CHO Me Meso-1e	88	OHC, Me	99	1:1
OHC Me	92	OHC,,2f	99	2:1

lization. This effect was demonstrated with 4-methyl-substituted 1d, which provided all four possible diastereomeric aldols of 2d upon treatment with proline; the resulting aldols were obtained in a 22:5:5:1 ratio and in 75, 89, > 95, and 8% ee, respectively. Explaining the lowered stereoselectivity in this case is difficult without calculating the relative energies of all reasonable transition states.^[12] Even if the enamine is fixed to having E geometry, and boat conformations are excluded (as in **B**), the transition states may still vary in an axial versus equatorial 4-substituent and/or enamine double bond, and in the anti versus syn relationship of the carboxylic acid to the olefin. We also investigated the behavior of the mesoconfigured dialdehyde 1e under our reaction conditions. Four stereogenic centers may be created simultaneously in a catalytic asymmetric desymmetrization of this substrate. We found the two expected anti-configured aldols (2e and 2e') to be formed in equal amounts and in 99 and 75% ee, respectively. That the proline-catalyzed enolexo aldolization is not limited to dialdehydes was illustrated in the reaction of ketoaldehyde 1 f, which gave tertiary aldol 2 f (d.r. = 2:1), in 99% ee (minor isomer 95% ee).

In summary, we have described the first and highly enantioselective proline-catalyzed *enolexo* aldolization of dicarbonyl compounds. This reaction provides β -hydroxy cyclohexane carbonyl derivatives that are of potential widespread usage in target-oriented synthesis. This *anti*-diastereoselective proline-catalyzed *enolexo* aldolization nicely complements alternative methodologies such as the highly enantio- and *syn*-diastereoselective baker's yeast reduction of β -keto esters. An advantage of the aldolization methodology is that both enantiomeric products can be accessed simply by using either (S)- or (R)-proline, whereas the biocatalysis route is limited to products of a single absolute configuration. Applications in natural product synthesis and further extensions of proline-catalyzed inter- and intramolecular aldolizations are forthcoming.

Experimental Section

Typical aldolization procedure: Dicarbonyl 1 (1 mmol) was dissolved in dry dichloromethane (10 mL) and treated with (S)- or (R)-proline (12 mg, 0.1 mmol, 10%). The mixture was stirred at room temperature until the starting material had disappeared (S-16 h). Aldols 2 can be isolated after standard aqueous work-up, but are unstable over extended time periods at room temperature. Stable diols are obtained by in situ reduction with NaBH₄ followed by an aqueous work-up, as described elesewhere. [11d,21]

Received: February 24, 2003 [Z51266]

Keywords: aldol reaction · amino acids · asymmetric catalysis · organocatalysis

For some recent reviews and accounts, see: a) J. S. Johnson, D. A. Evans, Acc. Chem. Res. 2000, 33, 325-335; b) T. D. Machajewski, C.-H. Wong, Angew. Chem. 2000, 112, 1406-1430; Angew. Chem. Int. Ed. 2000, 39, 1352-1374; c) B. Alcaide, P. Almendros, Eur. J. Org. Chem. 2002, 1595-1601; d) S. E. Denmark, R. A. Stavenger, Acc. Chem. Res. 2000, 33, 432-440; e) S. G. Nelson, Tetrahedron: Asymmetry 1998, 9, 357-

- 389; f) M. Shibasaki, H. Sasai, T. Arai, T. Iida, *Pure Appl. Chem.* **1998**, *70*, 1027–1034; g) E. M. Carreira, R. A. Singer, *Drug Discovery Today* **1996**, *1*, 145–150.
- [2] a) Z. G. Hajos, D. R. Parrish, German Patent DE 2102623, 1971;
 b) U. Eder, G. R. Sauer, R. Wiechert, German Patent DE 2014757, 1971;
 c) U. Eder, G. Sauer, R. Wiechert, Angew. Chem. 1971, 83, 492-493; Angew. Chem. Int. Ed. Engl. 1971, 10, 496-497;
 d) Z. G. Hajos, D. R. Parrish, J. Org. Chem. 1974, 39, 1615-1621;
 Related enantiogroup-differentiating aldol cyclodehydrations have been described, see: C. Agami, N. Platzer, H. Sevestre, Bull. Soc. Chim. Fr. 1987, 2, 358-360.
- [3] Reviews: a) N. Cohen, Acc. Chem. Res. 1976, 9, 512-517; b) K. Drauz, A. Kleemann, J. Martens, Angew. Chem. 1982, 94, 590-613; Angew. Chem. Int. Ed. Engl. 1982, 21, 584-608; c) E. R. Jarvo, S. J. Miller Tetrahedron 2002, 58, 2481-2495; d) B. List, Tetrahedron 2002, 58, 5572-5590; e) P. I. Dalko, L. Moisan, Angew. Chem. 2001, 113, 3840-3864; Angew. Chem. Int. Ed. 2001, 40, 3726-3748.
- [4] a) J. E. Baldwin, M. J. Lusch, *Tetrahedron* 1982, 38, 2939 2947;
 b) J. E. Baldwin, *J. Chem. Soc. Chem. Commun.* 1976, 734 736.
- [5] Cycloaldolizations are formally enolexo- or enolendo-exo-trig processes. However, since all intramolecular aldolizations are by definition exo-trig processes, we refer to these processes simply as enolexo or enolendo aldolizations.
- [6] For a proline-catalyzed *enolexo* aldolization in a dynamic kinetic resolution, see: R. B. Woodward et al., *J. Am. Chem. Soc.* 1981, 103, 3210–3213; Also see: C. Agami, N. Platzer, C. Puchot, H. Sevestre, *Tetrahedron* 1987, 43, 1091–1098.
- [7] For selected non-enantioselective variants, see: a) R. B. Woodward, F. Sondheimer, D. Taub, K. Heusler, W. M. MacLamore, J. Am. Chem. Soc. 1952, 74, 4223–4251; b) E. J. Corey, R. L. Danheiser, S. Chandrasekaran, P. Siret, G. E. Keck, J. L. Gras, J. Am. Chem. Soc. 1978, 100, 8031–8034; c) H. Hagiwara, H. Ono, N. Komatsubara, T. Hoshi, T. Suzuki, M. Ando, Tetrahedron Lett. 1999, 40, 6627–6630.
- [8] For two indirect catalytic enantioselective *enolexo* aldolizations, see: a) Romo's organocatalytic asymmetric *syn*-diastereospecific intramolecular, nucleophile-catalyzed aldol-lactonization (NCAL): G. S. Cortez, R. L. Tennyson, D. Romo, *J. Am. Chem. Soc.* 2001, 123, 7945–7946; b) Krische's catalytic asymmetric carbometallative aldol cycloreduction, which is a tandem catalytic asymmetric conjugate addition followed by a *syn*-diastereoselective *enolexo* aldolization: D. F. Cauble, J. D. Gipson, M. J. Krische, *J. Am. Chem. Soc.* 2003, 125, 1110–1111.
- [9] a) B. List, R. A. Lerner, C. F. Barbas III, J. Am. Chem. Soc. 2000, 122, 2395-2396; b) W. Notz, B. List, J. Am. Chem. Soc. 2000, 122, 7386-7387; c) B. List, P. Pojarliev, C. Castello, Org. Lett. 2001, 3, 573-575.
- [10] a) A. B. Northrup, D. W. C. MacMillan, J. Am. Chem. Soc. 2002, 124, 6798-6799 (This important paper describes an intermolecular variant of the reaction discussed here); b) A. Córdova, W. Notz, C. F. Barbas III, J. Org. Chem. 2002, 67, 301-303; c) A. Bøgevig, N. Kumaragurubaran, K. A. Jørgensen, Chem. Commun. 2002, 620-621, Also see: A. Bøgevig, K. Juhl, N. Kumaragurubaran, W. Zhuang, K. A. Jørgensen, Angew. Chem. 2002, 114, 1868-1871; Angew. Chem. Int. Ed. 2002, 41, 1790-1793; N. Halland, P. S. Aburel, K. A. Jørgensen, Angew. Chem. 2003, 115, 685-689; Angew. Chem. Int. Ed. 2003, 42, 661-665.
- [11] For the first proline-catalyzed asymmetric intermolecular Mannich, Michael, and α-amination reactions, see a) B. List, *J. Am. Chem. Soc.* 2000, 122, 9336–9337; b) B. List, P. Pojarliev, W. T. Biller, H. J. Martin, *J. Am. Chem. Soc.* 2002, 124, 827–833; c) B. List, P. Pojarliev, H. J. Martin, *Org. Lett.* 2001, 3, 2423–2425; d) B. List, *J. Am. Chem. Soc.* 2002, 124, 5656–5657. For recent highlights, see: a) M. Movassaghi, E. N. Jacobsen, *Science* 2003, 298, 1904–1905; b) S. Borman, *Chem. Eng. News* 2002, 80, 35–

Zuschriften

- 37; c) H. Gröger, J. Wilken, *Angew. Chem.* **2001**, *113*, 545–548; *Angew. Chem. Int. Ed.* **2001**, *40*, 529–532.
- [12] a) S. Bahmanyar, K. N. Houk, J. Am. Chem. Soc. 2001, 123, 9922-9923; b) S. Bahmanyar, K. N. Houk, J. Am. Chem. Soc. 2001, 123, 11273-11283; c) S. Bahmanyar, K. N. Houk, H. J. Martin, B. List, J. Am. Chem. Soc. 2003, 125, 2475-2479; d) L. Hoang, S. Bahmanyar, K. N. Houk, B. List, J. Am. Chem. Soc. 2003, 125, 16-17.
- [13] For additional supporting density functional theory calculations, see: a) K. N. Rankin, J. W. Gauld, R. J. Boyd, J. Phys. Chem. A 2002, 106, 5155-5159; b) M. Arnó, L. R. Domingo, Theor. Chem. Acc. 2002, 108, 232-239.
- [14] B. List, Synlett 2001, 1675-1686.
- [15] M. Daumas, Y. Vo-Quang, L. Vo-Quang, F. Le Goffic, Synthesis 1989, 64–65.
- [16] Comparison of the literature NMR data and optical rotation of the known diol of aldol 2a (T. Kakuchi, A. Namuri, H. Kaga, T. Ishibashi, M. Obata, K. Yokota, *Macromolecules* 2000, 33, 3964–3969) confirmed the expected absolute and relative configuration. The *ee* value of the aldol products was typically determined from chiral-phase HPLC after in situ conversion to a chromogenic enone by a Horner–Wadsworth–Emmons reaction (MeCOCH₂PO(OMe)₂, LiOH, THF).
- [17] The analogous proline-catalyzed 5-enolexo aldolization of hexanedial is less selective (d.r. = 2:1; ee(anti) = 79%, ee(syn) = 37%).
- [18] Substituted pimelaldehydes 1 were routinely prepared from commercially available substituted cyclohexanones through α-carbomethoxylation (NaH, MeOCO₂Me) and retro-Dieckman condensation using either NaOMe or Cs₂CO₃ (R. G. Salomon, M. F. Salomon, J. Org. Chem. 1975, 40, 1488–1492; P. Tundo, S. Memoli, M. Selva, , WO 02/14257, 2000.) Diisobutylaluminium hydride reduction (or LAH reduction followed by PCC oxidation) of the resulting diesters then provided the desired pimelaldehydes 1 in acceptable overall yields.
- [19] See for example: a) M. Bertau, M. Bürli, E. Hungerbühler, P. Wagner, *Tetrahedron: Asymmetry* 2001, 12, 2103–2107; b) V. Spiliotis, D. Papahatjis, N. Ragoussis, *Tetrahedron Lett.* 1990, 31, 1615–1616.
- [20] Alternative highly efficient and enantioselective ruthenium-catalyzed β-keto ester reductions can be diastereounselective: a) R. Noyori, T. Ohkuma, M. Kitamura, H. Takaya, N. Sayo, H. Kumobayashi, S. Akutagawa, J. Am. Chem. Soc. 1987, 109, 5856–5858; b) M. Kitamura, T. Ohkuma, M. Tokunaga, R. Noyori, Tetrahedron: Asymmetry 1990, 1, 1–4; c) J. P. Genêt, X. Pfister, V. Ratovelomanana-Vidal, C. Pinel, J. A. Laffitte, Tetrahedron Lett. 1994, 35, 4559–4562.
- [21] All new compounds gave satisfactory ¹H and ¹³C NMR spectra, as well as high-resolution mass spectroscopic analysis.